141. Syntheses of Novel Isopenam and Isocephem Antibiotics. Preparation of a Retinamido Derivative of a Highly Strained β -Lactam as **Potent Anticancer Agent**

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Syntheses of the cis-configurated isopenam **9** (Scheme *I),* isocephem **14** (Scheme 2), and isocephem **19** (Scheme **3)** are described. The key step in the preparation of **14** and **19** involved a *Pummerer-* type rearrangement of the corresponding sulfoxides **12** and **18.** These β -lactams were found to possess biological activity against several pathogenic microorganisms *in uitro.* The electronic activation of the lactam moiety of **19** remarkably enhanced its biological activity. A retinoic moiety was attached to **19** *via* an amino linker. The resultant retinamido-8-lactam **21** showed significant cytostatic activity in tracheal organ cultures obtained from vitamin-A-deficient hamsters.

Introduction. – Essential features of the classical β -lactam antibiotics penicillin **(I)** and cephalosporin **(II)** include *a*) a *cis*-fused β -lactam ring, *b*) an acylamino side chain, which can be varied considerably, c) an acidic function, and *d)* a five-membered ring or a six-membered ring containing a double bond conjugated with the β -lactam N-atom, thus conferring enough ring strain to raise the β -lactam frequency in the IR spectrum to ≥ 1765 cm⁻¹. The S-atom, however, can be replaced by an O- or a C-atom without substantial loss of antimicrobial activity [11. The IR absorption frequency of the carbonyl group of a β -lactam can also be considered as a measure of its reactivity towards nucleophilic attack *[2].* Therefore, higher frequency might indicate the potential for higher biological activity.

The syntheses of several monocyclic analogues of β -lactam antibiotics were reported [3] [4], in which the ring strain of fused β -lactams was replaced by an electron-withdrawing group (e, g, \textbf{III}) . Being susceptible to nucleophilic attack, β -lactam **III** does not exhibit antimicrobial activity. Therefore, the enamine fragment might have to be prevented from being coplanar with the β -lactam nucleus for biological activity. Because fused β -lactams

meet this requirement, we prepared isopenam **9** *(Scheme* I) and isocephem **14** *(Scheme* 2). We also report a synthesis of isocephem **19** *(Scheme 3)* in which the lactam moiety is activated electronically by an ester function. Furthermore, we attached a retinoic-acid moiety to **19** *via* an amino linker to afford compound **21** *(Scheme 4),* which exhibited anticancer activity.

Results and Discussion. - We treated trityl amine **(1)** with methyl glyoxylate **(2)** to produce the corresponding *Schijjf* base, which upon reaction with azidoacetyl chloride gave β -lactam 3 in 70% yield. The coupling constant (5.0 Hz) of the two H-atoms on the β -lactam ring indicated the *cis*-relationship of the two substituents [5] [6]. Reduction of the methoxycarbonyl group of **3** with NaBH, in wet THF gave alcohol **4** (60 *YO)* [7] [8], which was mesylated to afford **5** in 90% yield [9].

a) Dean-Stark trap, benzene. h) N,CH2COC1, Et,N. c) NaBH, wet THF. *d)* MeSO,CI, Et,N. e) KSCOCH,Ph, butanone. *f*) CF₃CO₂H, KClO₄. *g*) Br₂CHCO₂CH₂Ph, piperidine. *h*) Pd/C, EtOH.

A novel effect of KSAc was reported [lo] [ll], achieving the one-pot reduction/acylation of the azide function to the amide group. By this method and using KSCOCH,Ph, we obtained the cis-configurated phenylacetamide **6** in 40% yield from **5.** Removal of the trityl group from the N-atom of the β -lactam ring of 6 was more difficult than the corresponding detritylation **of** ordinary amides, amines, ethers, and esters [121 [13], probably because of the spatial arrangement of the lone-pair electrons of the β -lactam N-atom. We found that the addition of a trace of $KClO₄$ significantly accelerated the rate of deprotection of **6** with CF,COOH at *25",* giving **7** in 90 % yield after 1 h instead of 30 h; this rate enhancement may be attributed to the special salt effect that caused the rate of ionization of the trityl function equal the rate of product formation [14]. Monocyclic β -lactam 7 was then treated with benzyl dibromoacetate in the presence of piperidine in boiling t-BuOH to give **8** (45%). Hydrogenolysis **of 8** with PdjC in EtOH at 45 psi afforded isopenam **9** (60% yield).

Isocephem **14** was obtained from the key intermediate **7,** which was reacted with dibenzyl 2,3-dibromosuccinate **(10)** and piperidine in DMF to give a diastereoisomeric isocepham mixture **11** in 60% overall yield. Treatment of **11** with 3-chloroperbenzoic acid (3-C1C6H4C0,H) yielded the corresponding sulfoxide **12.** Reaction of **12** with AcCl and 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) afforded isocephem 13 (40% overall yield from **11).** In the presence of PdjC in EtOH at 45 psi, **13** was resistant to hydrogenolysis. By using PdCl, as the catalyst, we were able to achieve the reductive cleavage to isocephem **14** in 25% yield.

a) Piperidine/DMF, 80°. *b*) 3-ClC₆H₄CO₃H, CH₂Cl₂, 0°. *c*) AcCl, DBU. *d*) PdCl₂, AcOEt.

We started our synthesis of isocephem 19 with β -lactam 15 [4] [15] *(Scheme 3).* Reaction of **15** with bis(trimethy1tin) sulfide and Bu4NF afforded isocepham **16** as a mixture of two diastereoisomers in 80% overall yield. Hydrogenolysis of **16** with PdjC in MeOH containing 1% aq. NaHCO, solution at 60 psi followed by decarboxylation gave a diastereoisomeric mixture **17** of mono-carboxylic acids *(52%)* upon acidification with

a) (Me3Sn),S, Bu,NF. *b)* Pd/C, MeOH/1 % **aq.** NaHCO, **soh.,** 60 psi. c) 3-CIC6H4C03H, THF, *0". d)* **AcC1,** DBU.

AcOH. Treatment of 17 with 3-ClC₆H₄CO₃H gave the sulfoxide intermediate 18, which reacted with AcCl and DBU *in situ* to **19** through a *Pummerer-* type rearrangement. This one-pot conversion of **17** to isocephem **19** was accomplished in 50% overall yield.

Recently, we reported antileukemic effects of some azetidinone derivatives which contain a retinoic-acid chain such as **IV** [16] [17] (see *Scheme 4).* Retinoids exhibit

u) For 19:1. CICOOEt, pyridine; 2. H₂N(CH₂)₂NH₂. *b*) For 20: CICOOEt, pyridine.

anticancer properties for prophylaxis and can differentiate leukemic cells [18] [19]. On the other hand, β -lactams may act in a similar manner to alkylating agents to inhibit proliferation of the rapidly growing cells [20]. Therefore, we considered to combine a β -lactam with retinoic acid to give a new compound, which may have a special affinity for rapidly growing cells and thus may act as an effective anticancer agent. To this end, isocephem **19** was treated with ethyl chloroformate in pyridine to give the corresponding anhydride which upon reaction with ethylenediamine (amino linker) afforded an amide *(Scheme 4).* Retinoic acid **(20)** was also reacted with ethyl chloroformate in pyridine, and the resultant anhydride was added to the amide obtained from 19 to give retinamido- β lactam **21** in 85 % yield. Lactam **21** was stable under neutral conditions at 25-37" for one month.

Biological Activity. - Lactams **9, 14, 19,** and **21** as well as **IV,** ampicillin, cloxacillin, and penicillin G were tested *in vitro* against five pathogenic microorganisms up to level as high as 128 μ g/ml. The results are summarized in *Table 1*. The β -lactam moiety in compounds **I11** and **14** is activated electronically by a C=C-COOR functionality. In

Table 1. *Minimal Inhibitory Concentration* [kg/ml]

contrast to the freedom offered by the monocyclic β -lactam in **III**, the π -electrons of the C=C bond of the strained, bicyclic β -lactam 14 cannot not be aligned perfectly with the unshared electron pair of the N-atom. This discrepancy can account for the biological-activity difference between monocyclic and bicyclic p-lactams **111** and **14** (no antimicrobial activity of **111).** Results from our biological tests revealed the pronounced antimicrobial effect of β -lactams **9** and **14**. Consequently, we conclude that the electronic activation of β -lactams, although important, is not enough to secure the antibacterial effect of the compounds. However, the profound antimicrobial effect of the highly strained isocephem **19**, with respect to **14**, indicates that the electronic activation of the β -lactam moiety by an electron-withdrawing group plays an important role in biological activity of bicyclic β -lactams.

We studied the carcinostatic property of retinoid **21,** which was evaluated according to its ability to inhibit squamous metaplasia and keratinization in organ cultures of trachea derived from vitamin-A-deficient hamsters [21] [22]. Organ cultures were used according to a method described previously [22]. We found that retinoid **21** was more active than β -retinoic acid (20) and **IV** (*Table 2*). The potent carcinostatic property of retinoid 21 might also reflect the ring strain of β -lactam.

Retinoid ED_{50}° [M]	$1.0 \cdot 10^{-11}$	$7.8 \cdot 10^{-11}$	β -Retinoic acid (20) $1.15 \cdot 10^{-10}$
			Mean effective doses (ED_{50}) for the reversal of keratinization of 50% of the explants, as determined by probit

Table 2. *Activity of Retinoids in Tracheal Organ Cultures Obtained from Vitamin-A-Deficient Hamsters*

analysis [23] [24].

Furthermore, we determined the LD_{50} of the biologically active compounds in rats: thus, β -lactam **19** was administered at different doses intravenously (i, v) and retinoid 21 subcutaneously **(s.c.).** Compounds **19** and **21** did not show any toxicity at concentration levels as high as 500 and 100 mg/kg, respectively. **All** rats were controlled and were in good conditions after **6** months of administration. Nevertheless, an *LD,, (i.0.)* of *ca.* 800 mg/kg was determined for isocephem **19** and an *LD,,* **(s.c.)** of 160 mg/kg for retinamido- β -lactam **21**.

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Experimental Part

General. Chemicals were purchased from *Fluka Chemical Co.* Reagent-grade solvents were distilled and then stored over molecular sieves (4 A). Products were isolated by column chromatography (CC; *Merck* silica gel 60 (230-400 mesh), packed in glass column (20 g of silica gel/g of crude material)). TLC: *Merck* silica gel **60** F254 anal. sheets. **M.p.:** *Buchi 510.* IR Spectra: *Beckman-IR-8* spectrophotometer. 'H-NMR Spectra: *Bruker- WH-90, Varian-XL-200,* and *Varian- T-60A* spectrometers.

Methyl cis-3-Azido-4-oxo-1-tritylazetidine-2-carboxylate (3). To a soln. of trityl amine (1; 2.28 g, 0.01 mol) in dry benzene (350 ml) was added methyl glyoxylate **(2;** 4.5 g, 0.05 mol) in portions within **8** h at 80" *(Dean-Stark* trap). After H₂O was all removed (ca. 10 h), the soln. was cooled, anh. MgSO₄ added, the mixture filtered after 2 h, and the filtrate evaporated. Then dry CH_2Cl_2 (150 ml) and Et_3N (2.02 g, 0.02 mol) were added, followed by dropwise addition of azidoacetyl chloride (1.20 **g,** 0.01 mol) at reflux temp. After stirring at the same temp. for 7 h, the soln. was washed with H_2O , dried (MgSO₄), and evaporated. The crude product was purified by CC (silica gel, CH₂Cl₂): **3** (2.7 g, 70%). Oil. IR (CH₂Cl₂): 2100 (N₃), 1778 (β -lactam), 1750 (ester). ¹H-NMR (CDCl₃): 3.81 (s, Me); 4.51 *(d, J* = 5, H-C(3)); 4.89 *(d, J* = 5, H-C(4)); 7.23 (br. **s,** 3 Ph). Anal. calc. for C,,H,,N,O, (412.22): C 69.90, H 4.85, N 13.59; found: C 69.83, H 4.81, N 13.50.

cis-3-Azido-4- *(hydroxymethyl)-l-tritylazetidin-2-one* **(4)** was obtained from **3** in 60 % yield as described in [7] [8]. IR (CH2CI,): 3300-3370 (OH), 2100 (N,), 1770 (p-lactam). 'H-NMR (CDCI,): 3.25-3.50 *(m,* CH20H); 3.71-4.20 *(m, H-C(4))*; 4.43 *(d, J = 5, H-C(3))*; 7.31 (br. *s*, 3 Ph). Anal. calc. for C₂₃H₂₀N₄O₂ (384.12): C 71.87, H 5.21, N 14.58; found: C 71.80, H 5.11, N 14.60.

cis-3-Azido-4-[(mesyloxy)methyl]-l-tritylazetidin-2-one (5) was prepared from **4** in YO% yield according to [9]. IR (CH₂Cl₂): 2100 (N₃), 1777 (β -lactam). ¹H-NMR (CDCl₃): 2.80 (s, MeSO₃); 3.66 (br., MsOCH₂); 4.10-4.33 *(m, H-C(4)); 4.49 (d, J = 5, H-C(3)); 7.20 (s, 3 Ph). Anal. calc. for C₂₄H₂₂N₄O₄S (462.32): C 62.34, H 4.76, N* 12.12;found:C62.31,H4.77,N 12.21.

S-[*cis-4-0~0-3-(phenylacetamido)-l-tritylazetidine-2-methyl] (Phenyl) thioacetate (6).* To a soln. **of** *5* (4.62 g, 0.01 mol) in butanone (100 ml) was added KSCOCH₂Ph (7.64 g, 0.04 mol). The mixture was heated at reflux for 24 h, the solvent evaporated, and the resultant red-brown syrup taken up in AcOEt and washed with H₂O ($3\times$). The org. layer was dried **(MgSO,,),** filtered, and evaporated. CC (silica gel, CHCI,) gave *6* (2.4 g, 40%). Foam. 1R $(CH_2Cl_2): 3410$ (NH), 1780 (β -lactam), 1730 (thioester), 1681 (amide). ¹H-NMR (CDCl₃): 3.11–3.33 (m, CH₂S); 3.39 (s, CH₂COS); 3.43 (s, CH₂CO); 3.90–4.21 *(m, H*–C(2)); 5.27 *(dd, J* = 5, 9, H–C(3)); 6.81 (br., NH); 6.82–7.75 *(m, 5 Ph).* Anal. calc. for C₃₉H₃₄N₂O₃S (610.42): C 76.72, H 5.57, N 4.59; found: C 76.71, H 5.52, N 4.60.

S-[cis-4-Oxo-3-(phenylacetamido)azetidine-2-methyl] (Pheny1)thioacetate (7). To a soln. of *6* (3.0 g, 0.005 mol) in CF₃COOH (30 ml), a trace amount of KClO₄ was added and the soln. stirred at 25° for 1 h. Evaporation and CC (silica gel, AcOEt) gave 1.60 g (90%) of 7. Foam. IR (CH₂Cl₂): 1763 (β -lactam), 1720 (thioester), 1680 (amide). ¹H-NMR (CDCl₃): 3.01–3.42 *(m, CH*₂S); 3.41 *(s, CH*₂COS); 3.42 *(s, CH*₂CO); 4.10–4.31 *(m, H*–C(2)); 5.32 *(dd, J* = 4.5, 9, H–C(3)); 6.70–6.82 (br., 2 NH); 7.20, 7.31 (2s, 2 Ph). Anal. calc. for C₂₀H₂₀N₂O₃S (368.32): C 65.22, H 5.43, N 7.61 ; found: C 65.19, H 5.43, N 7.57.

Benzyl (2RS,5SR,6SR)-7-0xo-6-(phenylacetamido)-3-thia-l-azabicycIo[3.2.0]heptane-2-eurboxylute (8). To a soln. of **7** (3.68 g, 0.01 mol) in t-BuOH (50 ml) containing benzyl dibromoacetate (3.1 g, 0.01 mol), piperidine (8.50 g, 0.1 mol) was added and the soln. heated at reflux for 2 h. The solvent was evaporated under reduced pressure and the residue taken up in AcOEt and washed with 1% aq. HCl soln. (2x) and H_2O . The org. layer was dried (MgSO₄), filtered, and evaporated. CC (silica gel, CHCl₃) afforded **8** (1.80 g, 45%). Foam. IR (CH₂Cl₂): 3415 (NH); 1783 (β -lactam), 1742 (ester), 1670 (amide). ¹H-NMR (CDCl₃): 2.77–3.21 (*m*, CH₂S); 3.52 (s, CH₂CO); 4.32–4.58 *(m,* H–C(5)); 5.01 (s, CHCOO); 5.20 (s, CH₂O); 5.23–5.59 *(dd, J* = 5, 9.5, H–C(6)); 6.80–7.00 (br., NH); 7.31 (br. s, 2 Ph). Anal. calc. for C₂₁H₂₀N₂O₄S (396.22): C 63.63, H 5.05, N 7.07; found: C 63.60, H 5.13, N 7.13.

(2RS,5SR,6SR/-7-0xo-6-(phenylacetamidoj-3-thia-I-azabicyclo[3.2.0]heptane-2-carboxylic Acd(9). **A** mixture of (1.32 g, 0.0033 mol), EtOH (60 ml), and 10% Pd/C (20 mg) was hydrogenated at 25° and 45 psi for 2 h, during which the pressure dropped to 41 psi. The soln. was then filtered and evaporated to give 0.70 g (70 %) of **9,** m.p. 122-126". CC (silica gel, AcOEt) gave 60% of **9.** M.p. 141-143". IR (nujol): 3300-3500 (NH, COOH), 1779 (B-lactam), 1705 (acid), 1669 (amide). 'H-NMR ((D6)DMSO): 2.72-3.12 *(m,* CH,S); 3.50 **(s,** CH,CO); 4.214.28 *(m,* H-C(5)); 4.88 **(s.** CHCOO); 5.02-5.31 *(dd, J* = 4.5, 9, H-C(6)); 6.91 (br., NH); 7.38 (s, Ph); 7.80-8.50 (br,, COOH). Anal. calc. for C₁₄H₁₄N₂O₄S (306.14): C 54.90, H 4.57, N 9.15; found: C 55.01, H 4.48, N 9.18.

Dibenzyl (6RS,7RS)-8-0xo-7-(phenylacetamido)-4-thia-l-azabicyclo[4.2.0]octane-2,3-dicarboxylute **(11).** To a soh. of **7** (3.68 g, 0.01 mmol) in DMF (40 ml), dibenzyl 2,3-dibromosuccinate **(10;** 4.56 g, 0.01 mol) and piperidine (8.5 g, 0.1 mol) were added. The soln. was heated at 80" for 3.5 h, then cooled, and after addition of AcOEt, the mixture was washed with H₂O (3x), dried (MgSO₄), and evaporated and the residue purified by CC $(CHCl₃): 3.26 g (60%) of 11 as an oily mixture of diastereoisomers. IR (CH₂Cl₂): 3421 (NH), 1770 (β -lactam), 1735$ (ester), 1680 (amide). 'H-NMR (CDCl,): 2.81-3.32 *(m.* CH2S); 3.50, 3.52 (2s, CH2CO); 3.71-3.95 *(m,* H-C(3)); 4.36-4.68 *(m, H-C(2), H-C(6))*; 5.01, 5.03, 5.20, 5.21 (4s, 2 CH₂O); 5.30-5.72 *(m, H-C(7))*; 6.85-7.23 (br., NH); 7.45 (br. s, 3 Ph). Anal. calc. for C₃₀H₂₈N₂O₆S (544.62): C 66.15, H 5.18, N 5.14; found: C 66.25, H 5.20, N 5.16.

Dibenzyl (6 RS.7RSI-8-Oxo- 7- *(phenylacetamido)-4-thiu-l-uzubicyclo[4.2.O]oct-2-ene-2,3-dicurboxylute* **(13).** At 0° , 3-ClC₆H₄CO₃H (1.72 g, 0.01 mol) was added to **11** (5.44 g, 0.01 mol) in dry CH₂Cl₂ (100 ml). After stirring at 0° for 1 h and at 25° for 30 min, 1% NaHCO₃ soln. (100 ml) was added, the org. layer dried (MgSO₄) and evaporated, and the crude sulfoxide 12 dissolved in dry CH₂Cl₂ (100 ml) at 0° and treated with 1,8-diazabicyclo[5.4.0]undec-7-ene (1.52 g, 0.01 mol) and AcCl(l.0 **g,** 0.013 mol). The soln. was stirred at *Oo* for 1 hand at reflux temp. for 2 h, then washed with 5% HCl, 5% NaHCO₃, and NaCl soln., dried (MgSO₄), and evaporated and the residue chromatographed (silica gel, CH₂Cl₂ (\rightarrow impurities), then CHCl₃): 2.0 g (40%) of **13.** Foam. IR (CH₂Cl₂): 3423 (NH), 1796 (B-lactam), 1750 (ester), 1730 (C=C), 1682 (amide). 'H-NMR (CDC1,): 2.85-3.40 *(m,* CH2S); 3.51 **(s,** CH2CO); 4.28 (br., H-C(6)); 5.10, 5.1 1 (2s, 2 CH,O); 5.12-5.45 *(dd, J* = 4.5, 9, H-C(7)); 6.82 (br., NH); 7.39, 7.25 (2s, 3 Ph). Anal. calc. for C₃₀H₂₆N₂O₆S (542.62): C 66.42, H 4.80, N 5.16; found: C 66.44, H 4.92, N 5.06.

(6RS,7RS)-8-0xo-7- *(phenylacetamido)-4-thia-l-azabicyclo[4.Z.O]oct-2-ene-2,3-dicurboxylute* **(14).** A mixture of **13** (2.71 g, 0.005 mol), AcOEt (200 ml) and PdCl₂ (600 mg) was hydrogenated at 25° and 40 psi for 4 h. After filtration and evaporation the crude foam was crystallized from Et,O: 0.45 g (25%) of **14.** M.p. 192-194". IR (nujol): 3150-3680 (2 COOH, NH), 1780 (B-lactam), 1710, 1700 (2 C=O, C=C), 1675 (amide). 'H-NMR ((D6)DMSO/D20): 2.39-2.76 *(m,* CH2S); 3.49 **(s,** CH2CO); 4.32 (br. *m,* H-C(6)); 5.01 *(d, ^J*= 5, CHND); 7.38 (s, Ph). Anal. calc. for $C_{16}H_{14}N_2O_6S$ (362.24): C 53.04, H 3.87, N 7.73; found: C 53.25, H 3.69, N 8.00.

2,2-Dibenzyl 3-Methyl (6RS,7RS)-8-Oxo-7-(phenylacetamido)-4-thia-1-azabicyclo[4.2.0]octane-2,2,3-tri*carboxylute* **(16).** To a soln. of **15** (7.60 g, 0.01 mol) and bis(trimethy1tin) sulfide (3.60 g, 0.01 mol) in dry THF (300 ml) at 0° was added dropwise, within 2 h, $Bu_4NF (0.02 \text{ mol})$ in dry THF (15 ml). The soln. was stirred at 25° for 1 h and then partitioned between Et₂O and H₂O, the Et₂O layer dried (MgSO₄) and evaporated, and the residue chromatographed (silica gel, CHCl₃/AcOEt 1:1): 4.8 g (80%) of **16**. Oil. IR (CH₂Cl₂): 3350 (NH), 1774 (β -lactam), 1730-1750 (ester), 1680 (amide). 'H-NMR (CDCl,): 2.70-3.42 *(m,* CH2S); 3.50 (br. **s,** CH,CO); 3.52, 3.53 (2s, MeO); 4.014.39 *(m,* H-C(3), H-C(6)); 5.07-5.30 *(m,* 2 CH20); 5.31-5.69 *(m,* H-C(7)); 7.00-7.46 *(m,* 3 Ph, NH). Anal. calc. for $C_{32}H_{30}N_2O_8S$ (602.54): C 63.79, H 4.98, N 4.65; found: C 63.68, H 4.96, N 4.67.

3-Methyl 2-Hydrogen (6RS,7RS)-8-0xo-7-(phenylacetamido)-4-thia-l-azabicyclo[4.2.0]octane-2,3-dicarboxylate (17). A soln. of 16 (3.01 g, 0.005 mol) in MeOH (100 ml) containing 1% aq. NaHCO₃ soln. (15 ml) was hydrogenated over 10% Pd/C (1.5 g) at 45° and 60 psi for 5 h. The mixture was filtered, and AcOH (20 ml) added, the solvent evaporated, and the residue purified by CC (silica gel, AcOEt/MeOH 9:l): 0.97 g (52%) of **17.** M.p. 130--133°. IR (nujol): 3200-3650 (COOH, NH), 1770 (β-lactam), 1730 (ester), 1700 (acid), 1668 (amide). ¹H-NMR

 $(CDC1_3/(D_6)DMSO/D_2O)$: 2.50-2.91 *(m, CH₂S*); 3.51 (br. s, CH₂CO); 3.56 (br. s, MeO); 4.03-4.50 *(m, H*-C(2), H-C(3), H-C(6)); 5.20–5.51 *(m, H-C(7))*; 7.41 *(s, Ph).* Anal. calc. for C₁₇H₁₈N₂O₆S (378.32): C 53.97, H 4.76, N 7.41; found: C 53.86, H 4.65, N 7.52.

3-Methyl 2-Hydrogen (6RS,7RS)-8-Oxo-7-(phenylacetamido)-4-thia-1-azabicyclo[4.2.0]oct-2-ene-2,3-dicar*boxylute* (19). At O", 3-CIC6H4C03H (1.72 g, 0.01 mol) was added to 17 (3.78 g 0.01 mol) in dry THF (100 ml). After 2 h of stirring (TLC: no starting materials left), **1,8-diazabicyclo[5.4.0]undec-7-ene** (3.04 g, 0.02 mol) and AcCl(3.0 g, 0.039 mol) were added. The soln. was stirred at 0° for 1 h and at reflux temperature for 4 h and then partitioned between AcOEt and H₂O. The org. layer was washed with 5% HCl soln. and H₂O, dried (MgSO₄), and evaporated. CC (silica gel, AcOEt/MeOH 9:1) gave 1.88 g (50%) of 19. M.p. 150-153°. IR (nujol): 3150-3655 (COOH, NH), 1790 @-lactam), 1750 (ester), 1710, 1703 (acid, C=C), 1675 (amide). 'H-NMR ((D,)DMSO/D,O): 2.45-3.05 *(m.* CH,S); 3.48 **(s,** CH,CO); 3.95 (s, MeO); 4.33 *(m.* H-C(6)); 5.03 *(d, J* = 5, CHND); 7.39 (s, Ph). Anal. calc. for $C_{17}H_{16}N_2O_6S$ (376.32): C 54.25, H 4.25, N 7.45; found: C 54.23, H 4.22, N 7.50.

Methyl (6RS,7RS)-8-Oxo-7-(phenylacetamido)-2-{[2-(retinamido) ethylamino]carbonyl}-4-thia-1-azabicy*clo[4.2.0]octa-2-ene-3-carboxylate* (21). To a suspension of 19 (0.376 g, 1.0 mmol) in dry CH,Cl, (10 ml) containing pyridine (0.20 g, 2.5 mmol), ethyl chloroformate (0.12 g, 1.1 mmol) was added dropwise under a stream of N_2 at -5° . The soln, was stirred for 15 min, then ethylenediamine $(0.060 \text{ g}, 1.0 \text{ mmol})$ was added through a syringe. Similarly, retinoic acid (20; 0.30 g, 1.0 mmol) in dry CH₂Cl₂ (10 ml) containing pyridine (0.2 g, 2.5 mmol) was treated with ethyl chloroformate (0.12 g, 1.1 mmol) under N_2 at -5° for 15 min. The resulting soln. was transferred by syringe to the above mixture obtained from 19. After 1 h, the soln. was washed with H₂O (3×), dried (Na₂SO₄), filtered, and evaporated. Crystallization from Et₂O/hexane 1:1 gave 0.60 g (85%) of 21. M.p. 71–74°. UV (EtOH): 350 (1897). **IR** (CH,Cl,): 3300-3420 **(3** NH), 1793 @-lactam), 1704 (ester), 1700 (C=C), 1680, 1645 (amides). ¹H-NMR (CDCl₃): 1.15 (s, Me₂C); 1.31-2.40 *(m, 3 Me, 3 CH₂)*; 2.61-3.10 *(m, CH₂S)*; 3.05 *(br. s,* NCH,CH,N); 3.58 (s, CH,CO); 3.90 *(s,* MeO); 4.20-5.11 *(m,* H-C(6), H-C(7)); 5.60-6.71 *(m,* 6 CH, 3 NH); 7.40 (s, Ph). Anal. calc. for $C_{39}H_{48}N_4O_6S$ (700.64): C 66.86, H 6.90, N 7.99; found: C 66.98, H 6.79, N 8.21.

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